

# Vacuum Ultraviolet (VUV) and Infrared (IR) Photo-Absorption Spectroscopy of Space-Related Ice: Ethanolamine under 1 keV Electron Irradiation

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# OUTLINE

**I. Background**

**II. Experimental Methodology**

**III. Results & Discussion**

**IV. Astrochemical Implications**

**V. Summary**

# Background

2 atoms	H <sub>2</sub>	AlF	AlCl	C <sub>2</sub> **	CH	CH <sub>+</sub>	CN	CO	CO <sub>+</sub>	CP	SiC	HCl	KCl	NH	NO	NS	NaCl	OH	PN	SO	SO <sup>+</sup>	SiN	SiO	SiS	CS	HF	HD	FeO ?	O <sub>2</sub>	CF <sub>+</sub>	SiH <sub>2</sub> ?	PO	AlO	OH <sub>+</sub>	CN <sub>-</sub>	SH <sub>+</sub>	SH	HCl <sub>+</sub>	TiO	ArH <sub>+</sub>	N <sub>2</sub>	NO <sub>+</sub> ?	NS <sub>+</sub>	HeH <sub>+</sub>	PO <sub>+</sub> (2021)				
3 atoms	C <sub>3</sub> *	C <sub>2</sub> H	C <sub>2</sub> O	C <sub>2</sub> S	CH <sub>2</sub>	H <sub>2</sub> CN	HCO	HCO <sub>+</sub>	HCS <sub>+</sub>	HOC <sub>+</sub>	H <sub>2</sub> O	H <sub>2</sub> S	HNC	HNO	MgC N	MgN C	N <sub>2</sub> H <sup>+</sup>	N <sub>2</sub> O	NaC N	OCS	SO <sub>2</sub>	C <sub>-</sub> SiC <sub>2</sub>	CO <sub>2</sub> *	NH <sub>2</sub>	H <sub>3</sub> + (*)	SiCN	AlNC	SiNC	HCP	CCP	AlOH	H <sub>2</sub> O <sub>+</sub>	H <sub>2</sub> Cl <sub>+</sub>	KCN	FeC N	HO <sub>2</sub>	TiO <sub>2</sub>	C <sub>2</sub> N	Si <sub>2</sub> C	HS <sub>2</sub>	HCS	HSC	NCO	CaN C	NCS (2021)				
4 atoms	C <sub>-</sub> C <sub>3</sub> H	C <sub>3</sub> H <sub>+</sub>	C <sub>3</sub> N	C <sub>3</sub> O	C <sub>3</sub> S	C <sub>2</sub> H <sub>2</sub> *	NH <sub>3</sub>	HCC N	HNC H <sub>+</sub>	HNC O	HNC S	HOC O <sub>+</sub>	H <sub>2</sub> C O	H <sub>2</sub> C N	H <sub>2</sub> CS	H <sub>3</sub> O <sub>+</sub>	C <sub>-</sub> SiC <sub>3</sub>	CH <sub>3</sub> *	C <sub>3</sub> N	PH <sub>3</sub>	H <sub>2</sub> CN O	HOC N	HSC N	H <sub>2</sub> O <sub>2</sub>	C <sub>3</sub> H <sub>+</sub>	H <sub>2</sub> Mg NC	HCC O	CNC N	HON O	MgC 2H	HCC S (2021)	HNC N (2021)	H <sub>2</sub> N C (2021)	HCC S <sub>+</sub> (2022)															
5 atoms	C <sub>5</sub> *	C <sub>4</sub> H	C <sub>4</sub> Si	C <sub>3</sub> H <sub>2</sub>	C <sub>3</sub> H <sub>2</sub>	H <sub>2</sub> CN	CH <sub>4</sub> *	HC <sub>3</sub> N	HCC NC	HCO OH	H <sub>2</sub> C NH	H <sub>2</sub> C 2O	H <sub>2</sub> N CN	HNC 3	SiH <sub>4</sub> *	H <sub>2</sub> C OH <sub>+</sub>	C <sub>4</sub> H <sub>-</sub>	H <sub>2</sub> C O	HNC NH	CH <sub>3</sub> O	NH <sub>4</sub> +	H <sub>2</sub> N CO <sub>+</sub>	NCC NH <sub>+</sub>	CH <sub>3</sub> Cl	MgC 3N	NH <sub>2</sub> OH	HC <sub>3</sub> O <sub>+</sub>	HC <sub>3</sub> S <sub>+</sub> (2021)	H <sub>2</sub> C S <sub>+</sub> (2021)	C <sub>4</sub> S (2021)	H <sub>2</sub> C O <sub>2</sub> SH (2021)	HC <sub>3</sub> S CN (2021)	HCC CO (2021)																
6 atoms	C <sub>5</sub> H	C <sub>4</sub> H <sub>2</sub>	C <sub>3</sub> H <sub>4</sub> *	CH <sub>3</sub> CN	CH <sub>3</sub> NC	CH <sub>3</sub> OH	CH <sub>3</sub> H	CH <sub>3</sub> NH <sub>+</sub>	HCC CHO	NH <sub>2</sub> CHO	C <sub>5</sub> N	C <sub>4</sub> H <sub>+</sub>	C <sub>4</sub> H <sub>-</sub>	C <sub>-</sub> H <sub>2</sub> C 3O	H <sub>2</sub> C CNH	C <sub>5</sub> N	HNC HCN	SiH <sub>3</sub> CN	C <sub>5</sub> S	MgC 4H	CH <sub>3</sub> CO <sub>+</sub>	C <sub>3</sub> H <sub>3</sub> (2021)	H <sub>2</sub> C 3S (2021)	HCC CHS (2021)	C <sub>5</sub> O (2021)	C <sub>5</sub> H + (2022)	HCC NCH + (2022)	C <sub>-</sub> C <sub>3</sub> C 2H (2022)																					
7 atoms	C <sub>6</sub> H	CH <sub>2</sub> CHC N	CH <sub>3</sub> C <sub>2</sub> H	HC <sub>5</sub>	CH <sub>3</sub> CHO	CH <sub>3</sub> NH <sub>2</sub>	C <sub>-</sub> C <sub>2</sub> H 4O	H <sub>2</sub> C CHO	C <sub>6</sub> H <sub>-</sub>	CH <sub>3</sub> NCO	HC <sub>5</sub> O	HOC H <sub>2</sub> C N	HCC CHN H	HC <sub>4</sub> NC	C <sub>-</sub> C <sub>3</sub> H CCH (2021)	H <sub>2</sub> C 5N (2021)	MgC 5N (2021)	CH <sub>2</sub> C <sub>3</sub> N (2021)																															
8 atoms	CH <sub>3</sub> C <sub>3</sub> N	HC O <sub>2</sub> O CH <sub>3</sub>	CH <sub>3</sub> COO C <sub>7</sub> H	C <sub>6</sub> H <sub>2</sub>	CH <sub>2</sub> OHC HO	C <sub>-</sub> HC <sub>6</sub> H*	CH <sub>2</sub> CHC HO	CH <sub>2</sub> CN	H <sub>2</sub> N CN	CH <sub>3</sub> CHN H	CH <sub>3</sub> S H <sub>3</sub>	H <sub>2</sub> N C(O) NH <sub>2</sub>	HCC CH <sub>2</sub> CN	HC <sub>5</sub> NH <sub>+</sub>	CH <sub>2</sub> CHC CH (2021)	MgC 6H (2021)	C <sub>2</sub> H 3NH <sub>2</sub> (2021)	ICH 2OH (2022)																															
9 atoms	CH <sub>3</sub> C <sub>4</sub> H	CH <sub>3</sub> CH <sub>2</sub> CN	ICH <sub>3</sub> I <sub>2</sub> O	CH <sub>3</sub> CH <sub>2</sub> OH	HC <sub>7</sub> N	C <sub>8</sub> H	CH <sub>3</sub> C(O) NH <sub>2</sub>	C <sub>8</sub> H <sub>-</sub>	C <sub>3</sub> H <sub>6</sub>	CH <sub>3</sub> CH <sub>2</sub> S NH <sub>2</sub>	CH <sub>3</sub> HO	HC <sub>7</sub> O	HCC CHC HCN (2021)	H <sub>2</sub> C CHC 3N (2021)	H <sub>2</sub> C CCH CCH (2021)	HOC HCH CHO (2022)																																	
10 atoms	CH <sub>3</sub> C <sub>5</sub> N	ICH <sub>3</sub> I <sub>2</sub> C O	ICH <sub>2</sub> OH <sub>2</sub>	CH <sub>3</sub> CH <sub>2</sub> CHO	CH <sub>3</sub> H <sub>2</sub> O	CH <sub>3</sub> OCH 2OH	C <sub>-</sub> C <sub>6</sub> H 4 (2021)	H <sub>2</sub> C CCH C <sub>3</sub> N (2021)	C <sub>2</sub> H 5NC O (2021)	C <sub>2</sub> H 5NH <sub>2</sub> (2021)	HC <sub>7</sub> NH <sub>+</sub> (2022)	CH <sub>3</sub> CHC HCN (2022)	CH <sub>3</sub> CIC NIC H <sub>2</sub> (2022)	CH <sub>2</sub> CHC H <sub>2</sub> C N (2022)																																			
11 atoms	HC <sub>9</sub> N	CH <sub>3</sub> C <sub>6</sub> H	C <sub>2</sub> H SOC HO	CH <sub>3</sub> OC O <sub>2</sub> C H <sub>3</sub>	CH <sub>3</sub> CH <sub>2</sub> OH	C <sub>-</sub> C <sub>5</sub> H 6 (2021)	HOC H <sub>2</sub> C H <sub>2</sub> N H <sub>2</sub> (2021)	H <sub>2</sub> C CCH C <sub>4</sub> H (2022)																																									
12 atoms	C <sub>6</sub> C <sub>6</sub> H 6*	n- C <sub>3</sub> H 7CN	i- C <sub>3</sub> H 7CN	C <sub>2</sub> H 5OC H <sub>3</sub>	C <sub>5</sub> H 5CN (2021)	2-c- C <sub>5</sub> H (2021)	CH <sub>3</sub> C <sub>7</sub> N (2022)	n- C <sub>3</sub> H 7OH (2022)	i- C <sub>3</sub> H 7OH (2022)																																								
>12 atoms	C <sub>60</sub> *	C <sub>70</sub> *	C <sub>60</sub> + C <sub>6</sub> H 5CN	HC <sub>1</sub> N	1- C <sub>10</sub> H <sub>7</sub> C N (2021)	2- C <sub>10</sub> H <sub>7</sub> C N (2021)	1-c- C <sub>9</sub> H <sub>5</sub> 8 (2021)	2-c- C <sub>5</sub> H 5CC H (2021)	C <sub>-</sub> C <sub>5</sub> H 4CC H <sub>2</sub> (2022)	2- C <sub>9</sub> H 7CN (2022)																																							

## MOLECULAR UNIVERSE

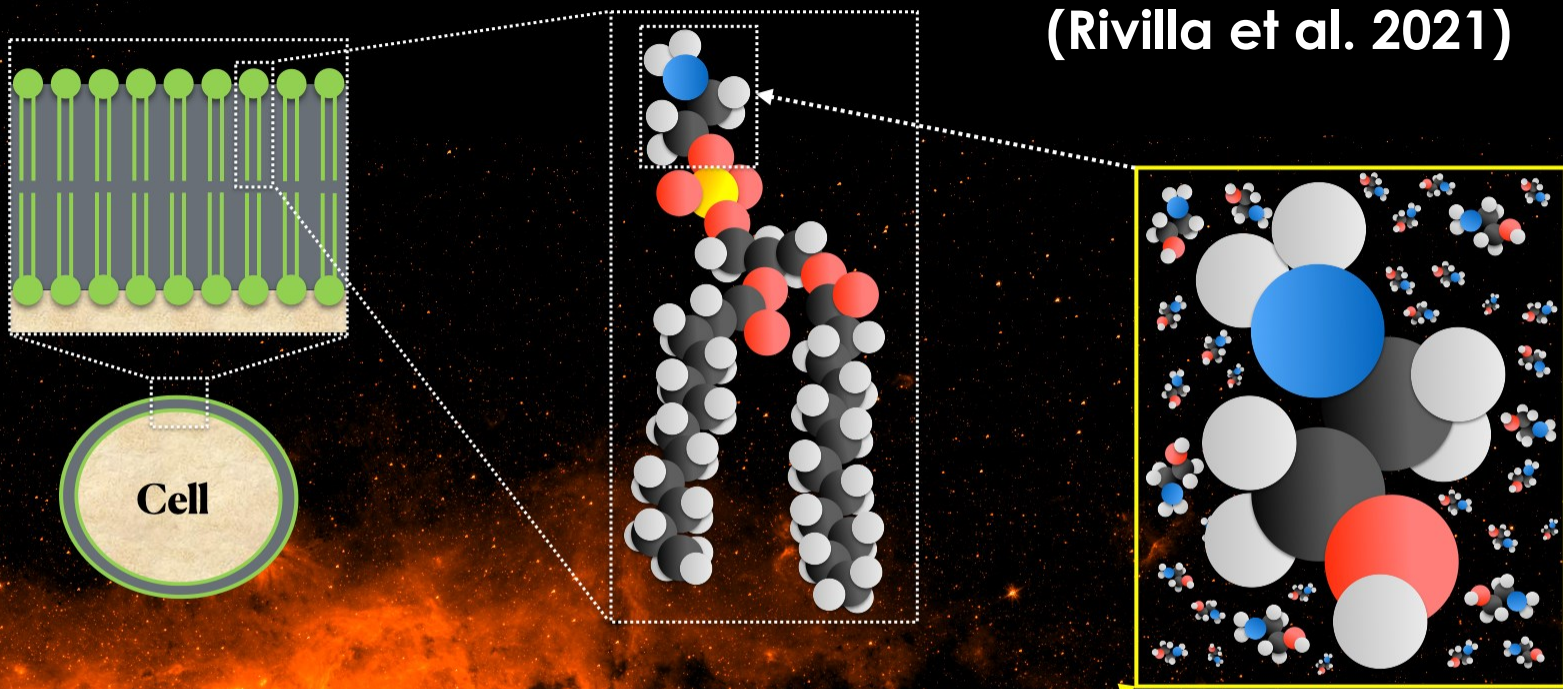
~ 270 molecules have been detected in the interstellar medium (ISM) or circumstellar shells as of Oct. 2022

One third being complex organic molecules (COMs)

➔ Ingredients of life in space!

<https://cdms.astro.uni-koeln.de/classic/molecules>

# Detection of Ethanolamine (EtA)



(Rivilla et al. 2021)

★  
Galactic Center

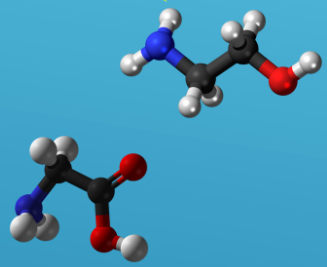
★

**G+0.693-0.027**  
**molecular cloud**

# Origin of Earth life



Delivery of prebiotic COMs from Space to Earth



**EtA in the ISM could be seeded to planetesimals during the formation of the Solar System and subsequently transferred to the early Earth.**

Existence of EtA in ISM and on the Almahata Sitta meteorite (in our Solar system),

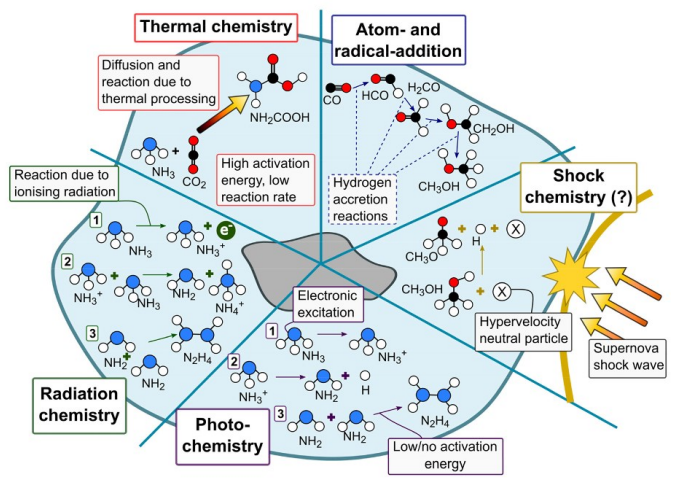


Efficient synthesis and stability under astrophysical conditions



# Stability of EtA in ISM

## Ice Grain Chemistry



Arumainayagam et al., *Chem. Soc. Rev.* (2019)

Cold clouds: molecules are in the solid phase on interstellar grains.

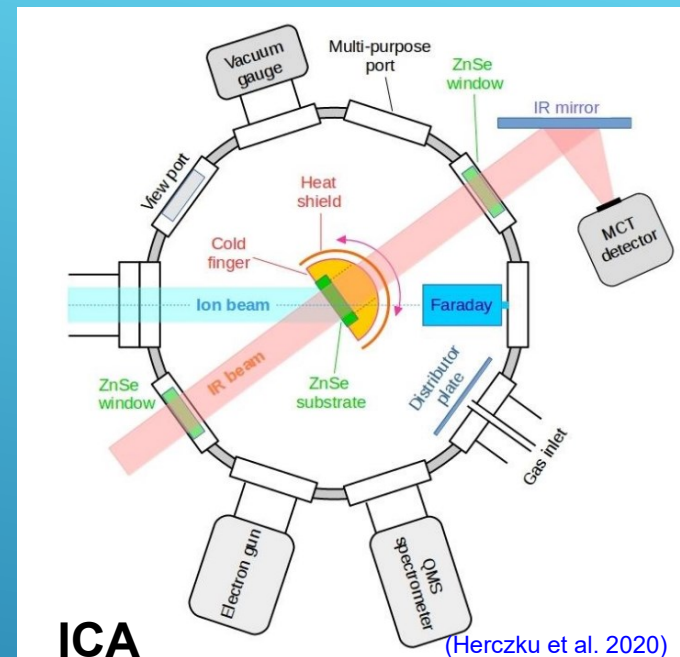
Chemical processes:

- Photochemistry
  - Radiation chemistry
  - Thermal chemistry
  - Shock chemistry
- }
- Energetic processing
- 
- Dark chemistry (surface atom addition reactions)
- }
- Non-energetic processing

- 1-keV electron beam irradiation to simulate the interaction of electron irradiation with interstellar icy mantles.

# Experimental methodology

## Mid-Infrared Spectroscopy Experiments at ATOMKI, Hungary

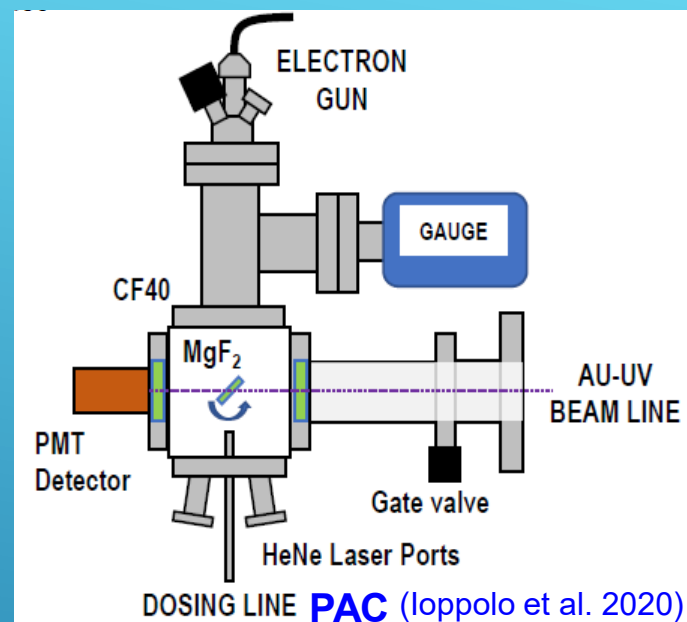


### Ice Chamber for Astrophysics-Astrochemistry (ICA):

- UHV-compatible chamber with a base pressure of  $10^{-9}$  mbar.
- ZnSe substrates cooled to 20 K.
- EtA ice was deposited at 20 K, and a mid-infrared absorption spectrum was acquired.
- Ice irradiated with 1 keV electrons.
- Mid-infrared absorption spectra were acquired throughout irradiation.

(spectral resolution =  $0.5 \text{ cm}^{-1}$ ;  
spectral range =  $4000\text{-}650 \text{ cm}^{-1}$ ;  
number of co-added scans =  
256 per spectrum).

# Vacuum-Ultraviolet (VUV) Spectroscopy at ASTRID-2 synchrotron, Aarhus University, Denmark



- The PAC was adjoined to the AU-UV beamline at the ASTRID-2 synchrotron light source (115-700 nm). A photomultiplier tube was connected to the PAC to detect the incident synchrotron radiation of the VUV spectroscopic beam.
- Degassed EtA was deposited, after which a VUV absorption spectrum was acquired.
- The EtA ice was then irradiated using 1 keV. Additional vacuum-ultraviolet absorption spectra were acquired throughout irradiation.

Open University Portable Astrochemistry Chamber (PAC): base pressure of  $10^{-9}$  mbar,  $MgF_2$  substrate, which may be cooled to 20 K.

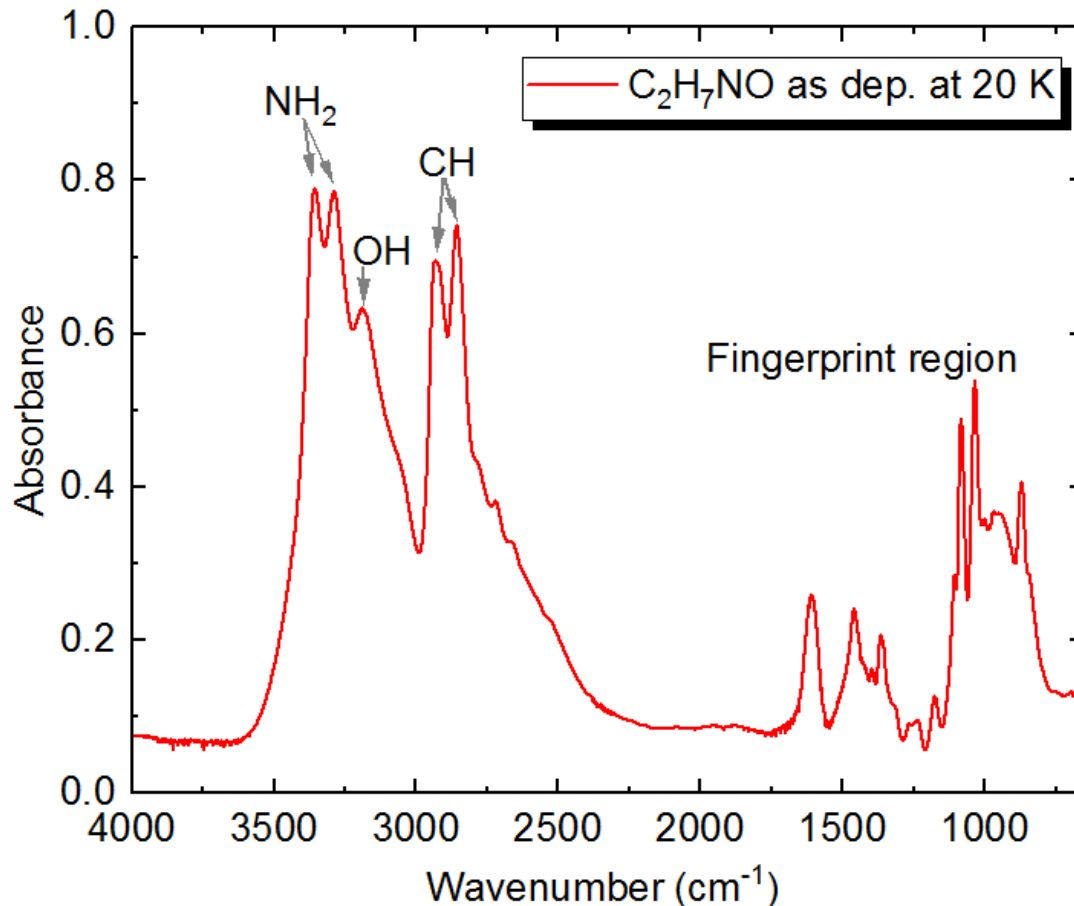
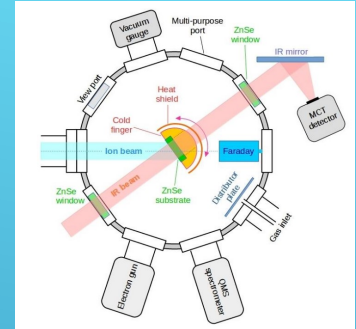
Electron irradiation with a flux of  $2 \times 10^{13}$  electrons $^{-1}$  cm $^{-2}$ , corresponding to an overall delivered fluence of  $7.2 \times 10^{16}$  electrons cm $^{-2}$



# Results & Discussion

## IR Absorption Spectrum of Deposited EtA ice

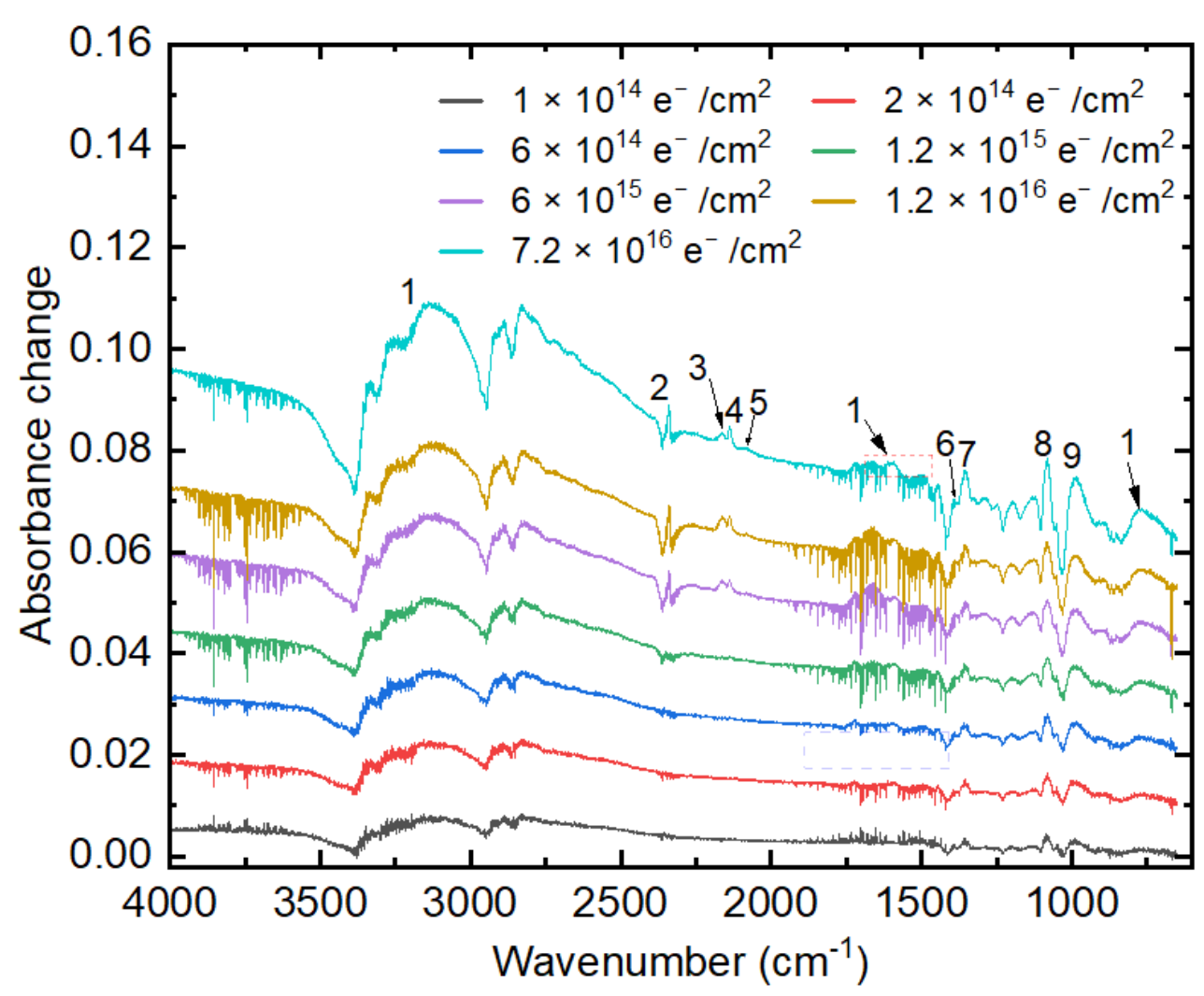
Spectral resolution:  $0.5 \text{ cm}^{-1}$



- Intensive peaks around  $3300 \text{ cm}^{-1}$  are assigned to the  $-\text{NH}_2$ , (symmetric and antisymmetric  $\text{NH}_2$  stretching fundamentals at  $3287$  and  $3353 \text{ cm}^{-1}$ ).
- OH stretching fundamentals at  $3187 \text{ cm}^{-1}$ ,
- CH stretching bands at  $2857$  and  $2933 \text{ cm}^{-1}$ .
- Fingerprint region (from about  $1500$  to  $650 \text{ cm}^{-1}$ ).

# Electron Irradiation: 1 keV e<sup>-</sup> on pure EtA

---To simulate secondary electrons emitted from ice mantles upon cosmic ray irradiation in cold clouds



**Spectra change during electron irradiation**

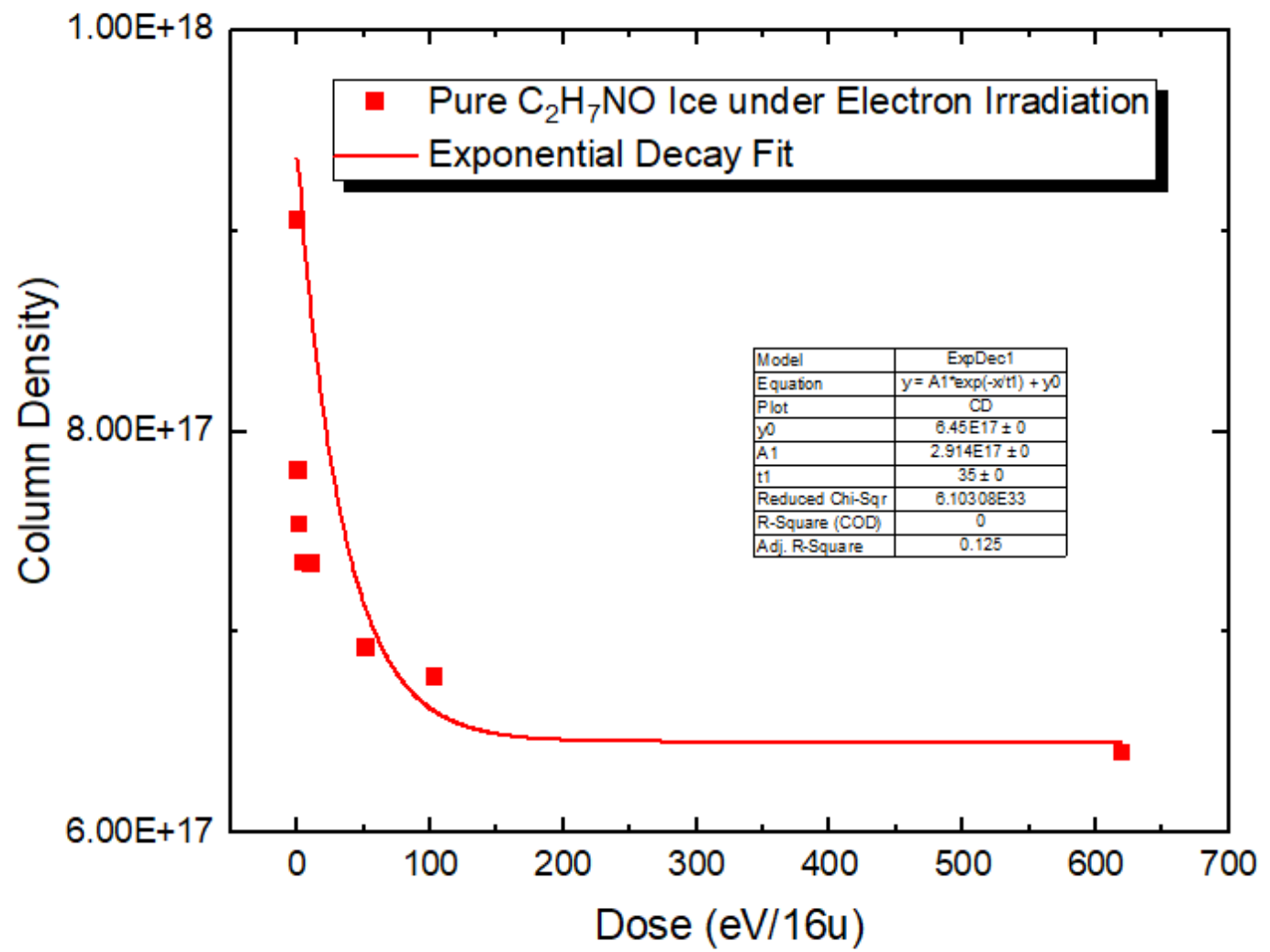
**Products formation from Destroyed EtA ice**

Band no.	1	2	3	4	5	6	7	8	9
Product	H <sub>2</sub> O	CO <sub>2</sub>	OCN <sup>-</sup>	CO	O <sub>3</sub>	C <sub>2</sub> H <sub>5</sub> OH	CH <sub>3</sub> CHO	NH <sub>3</sub>	C <sub>2</sub> H <sub>4</sub>

# Electron Irradiation: 1 keV e<sup>-</sup> on pure EtA

## Destruction of pure EtA ice under electron irradiation

Deposition Thickness: 885 nm

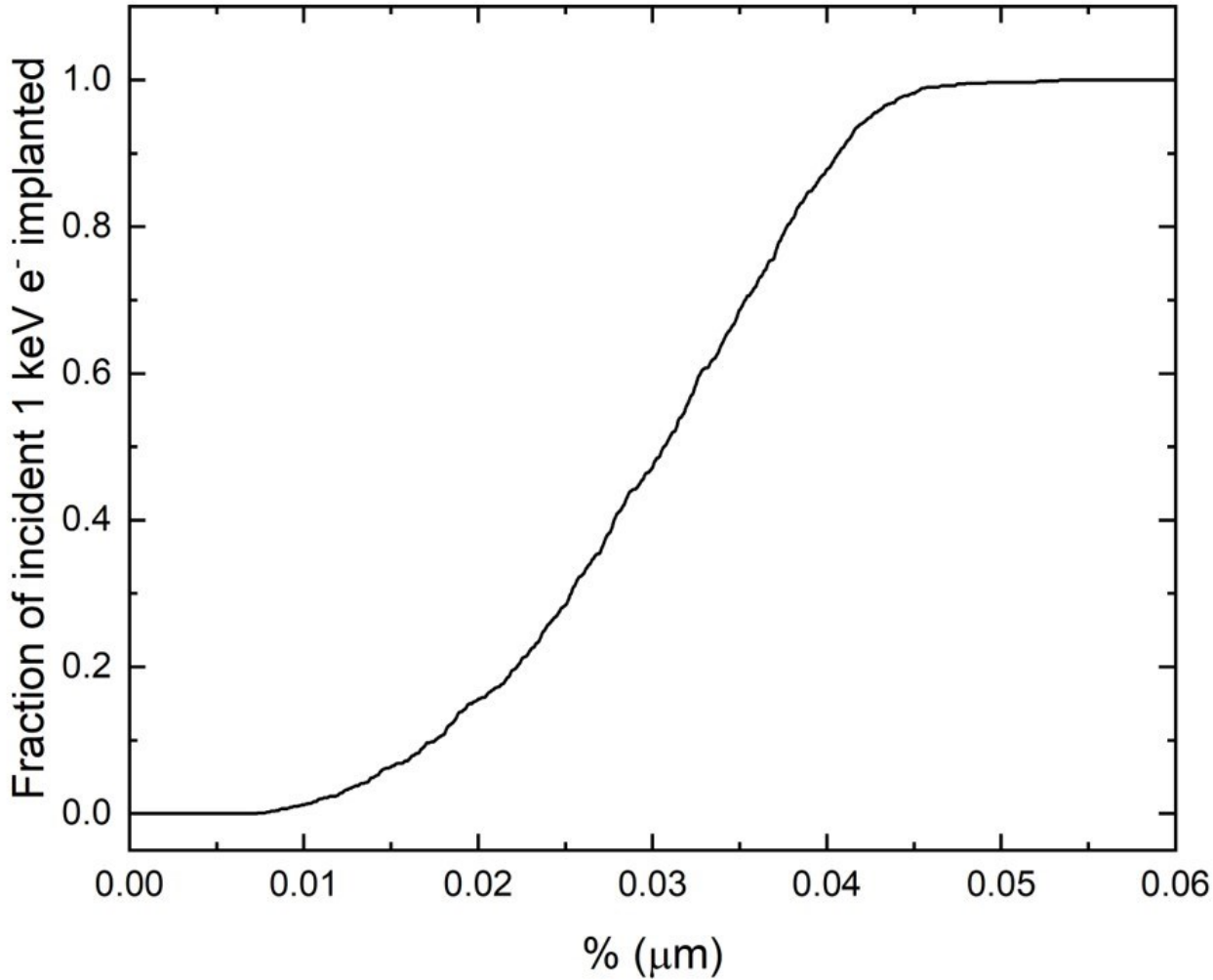


■ 29.3% EtA Destroyed

■ Half-life dose: 1.7 eV/16u

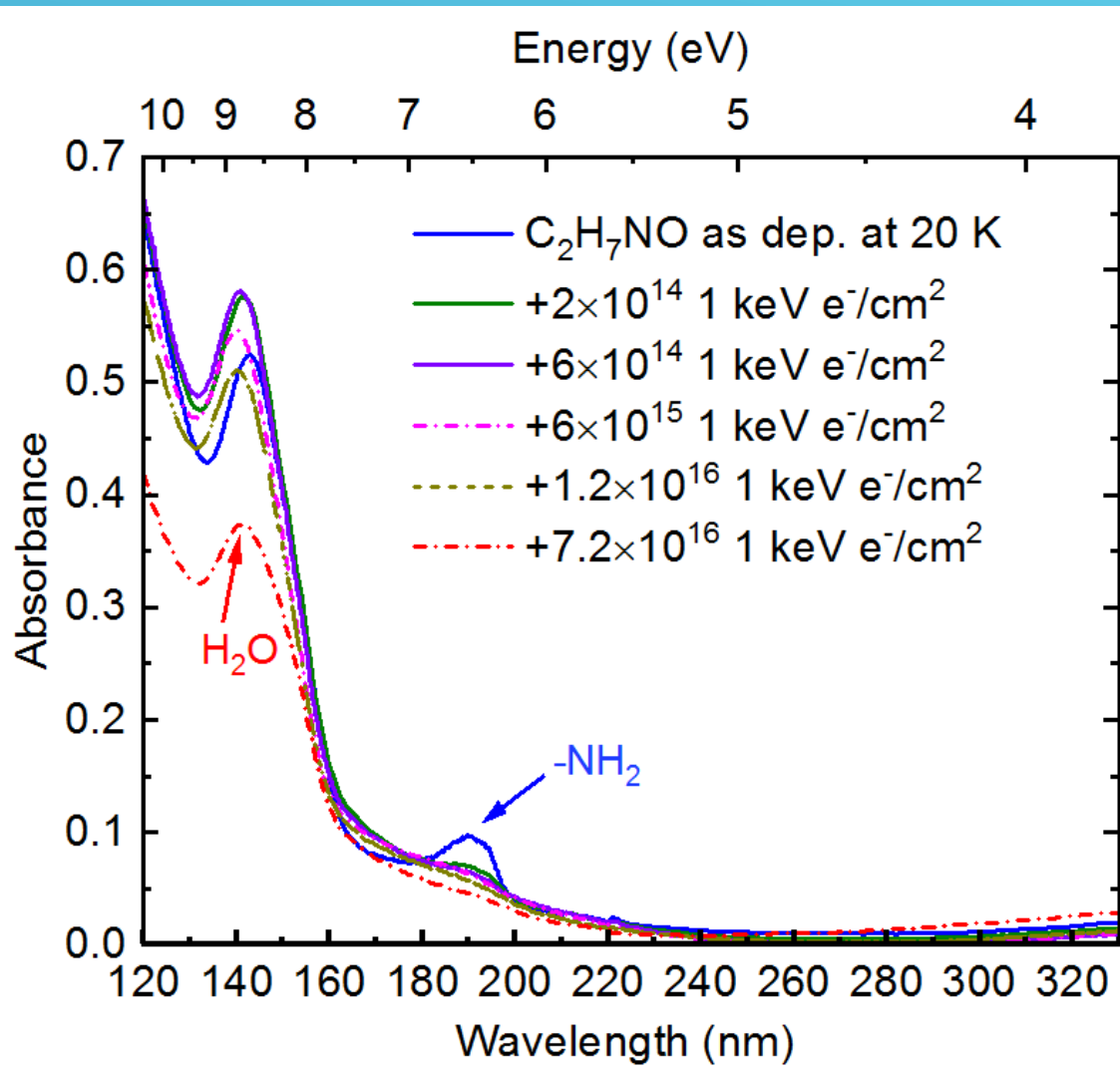
# Electron Irradiation: 1 keV e<sup>-</sup> on pure EtA

## CASINO simulation on electron implantation



**Penetration  
depth:  
45 nm**

# VUV Absorption Spectra of Deposited pure EtA ice



## Absorption features of deposited EtA ice

- A broad absorption peak centred at 143 nm.
- A minor band centred at 190 nm (amine,  $-NH_2$ ), assigned to the  $n \rightarrow \sigma^*$  transition of the lone pair of electrons in N atom.

## During irradiation

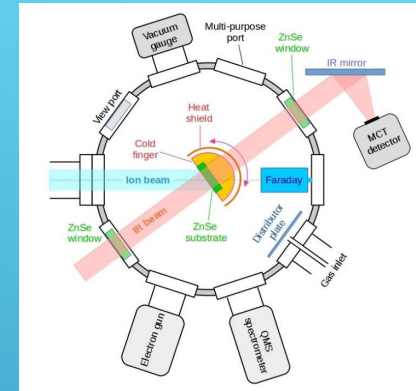
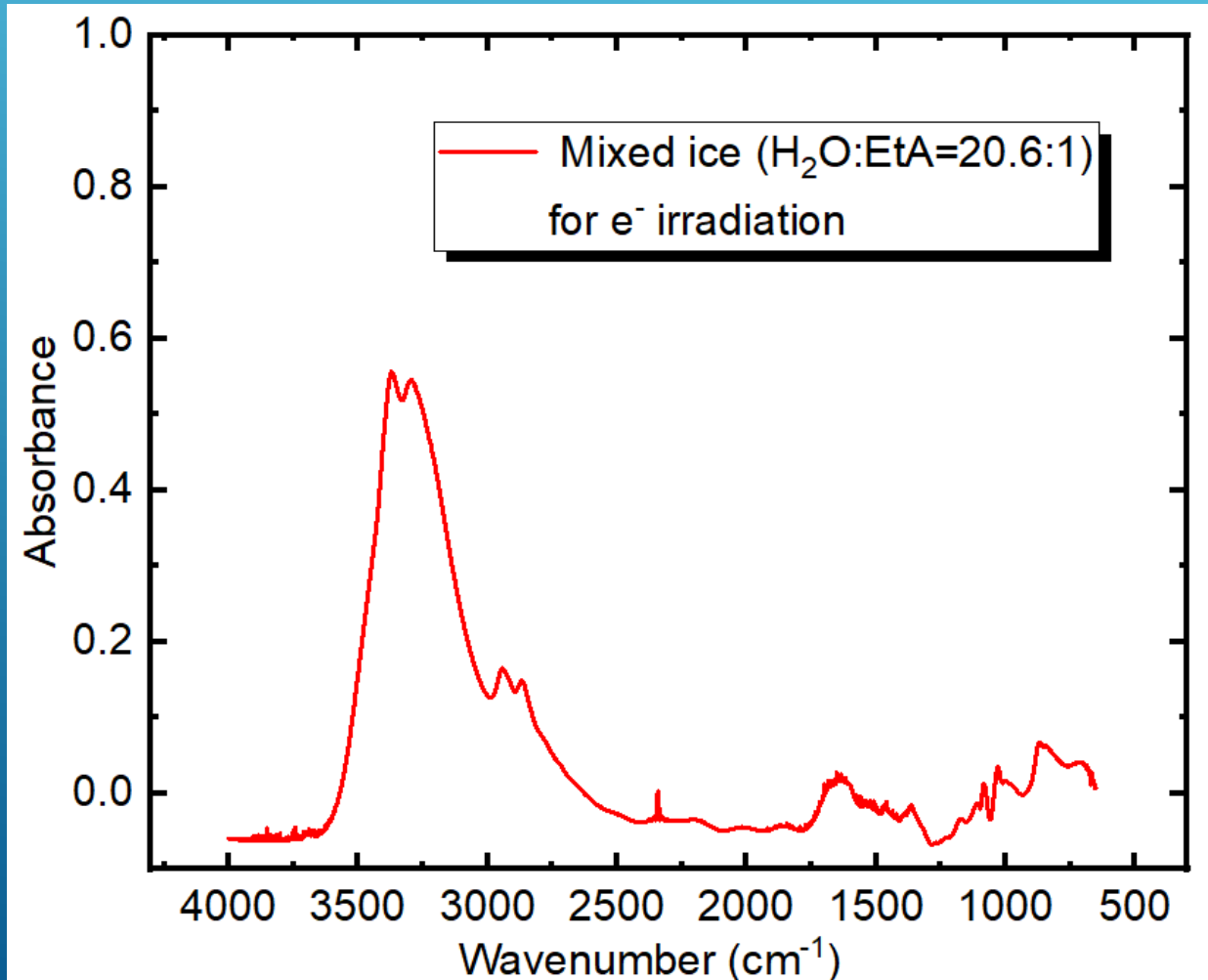
- The  $-NH_2$  peak disappears as irradiation dose grows, showing EtA was destroyed by the electrons, and almost all of the EtA was destroyed eventually.
- $H_2O$  was formed, with absorption peak at 142 nm (8.83 eV).

Deposition Thickness: 40 nm

# Electron Irradiation: 1 keV e<sup>-</sup> on mixed ice (H<sub>2</sub>O:EtA=20.6:1)

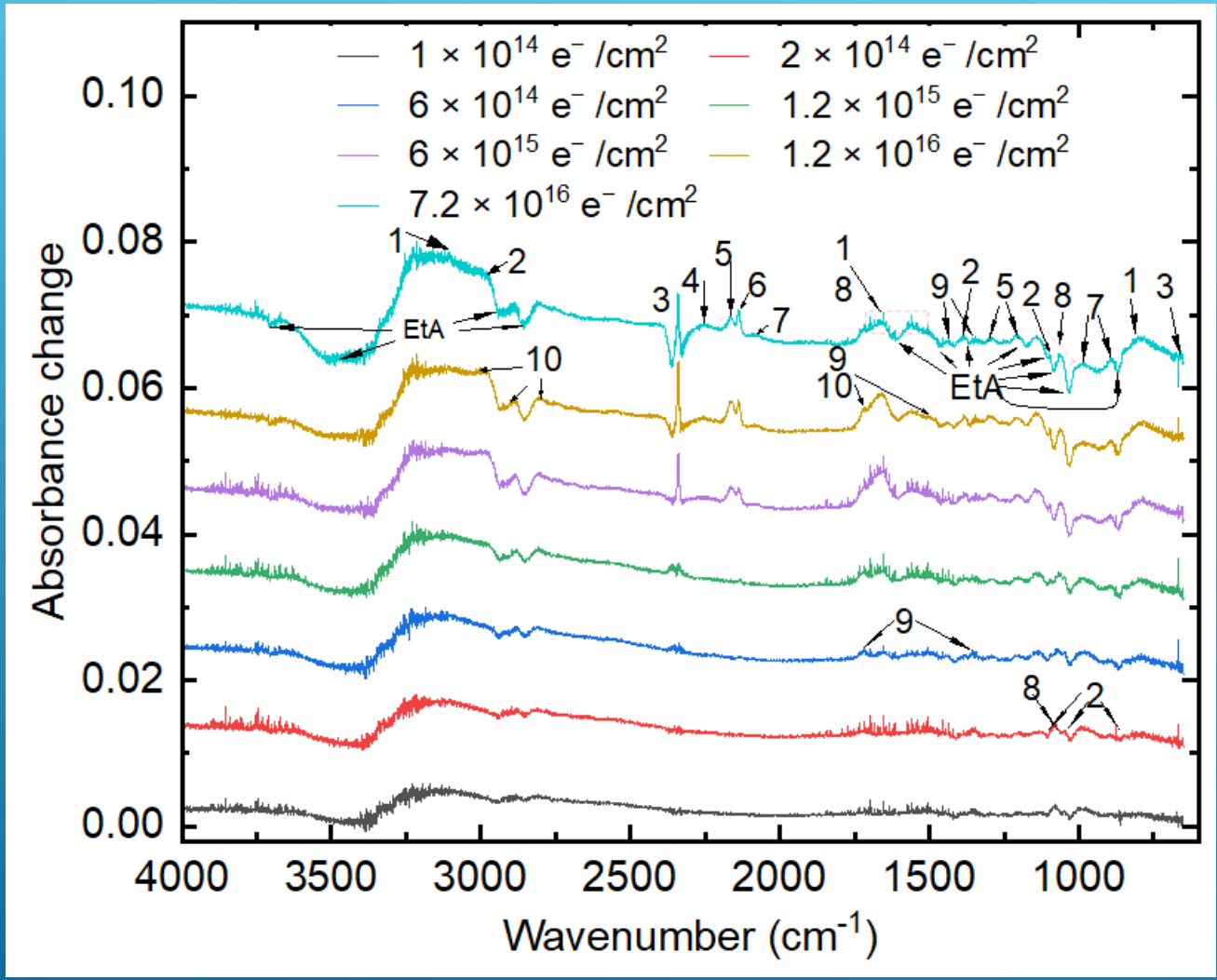
---To simulate EtA diluted in water under secondary electron irradiation in cold clouds

## IR absorption spectrum of deposited mixed ice



ICA (Herczku et al. 2020)

# Destruction of EtA and formation of products

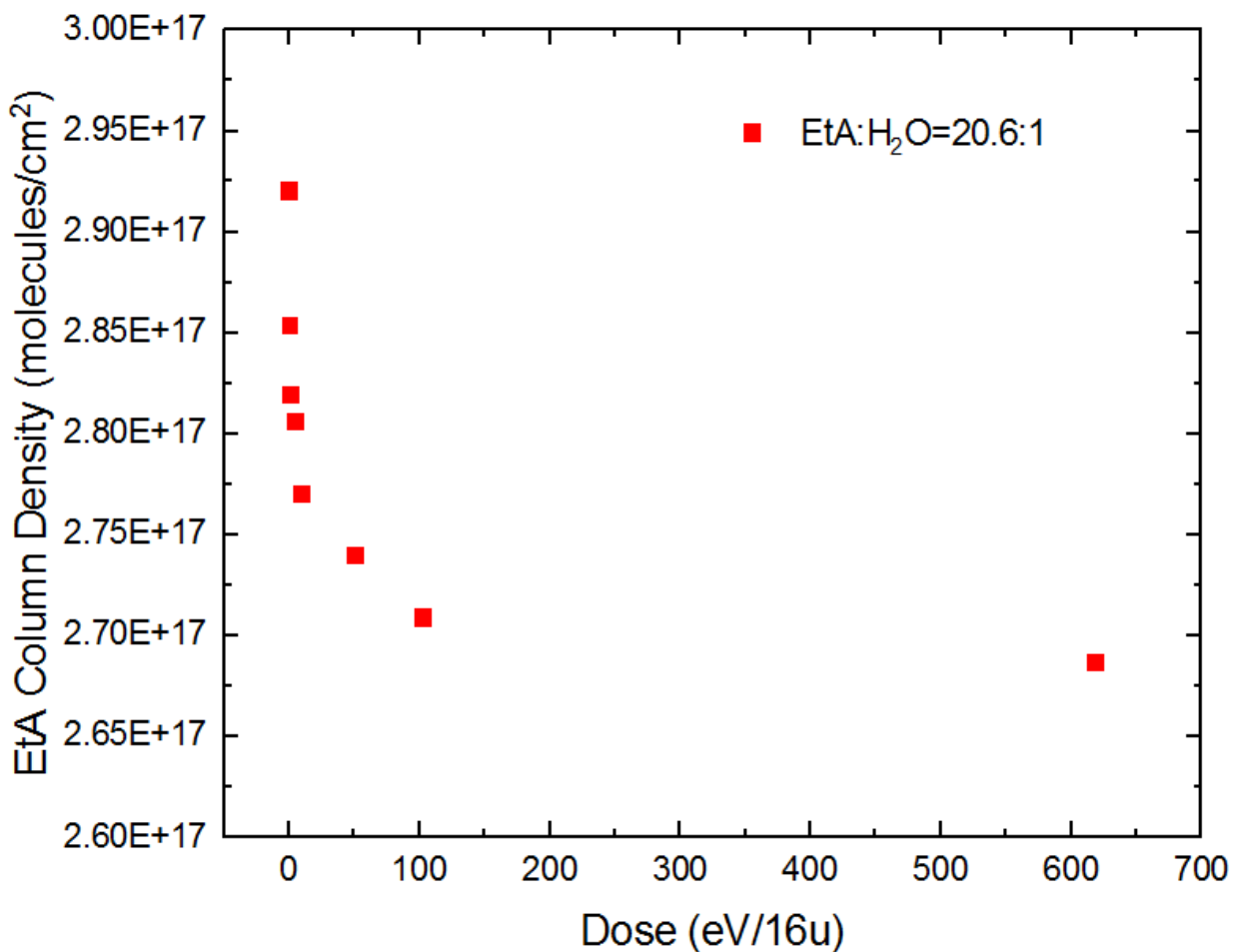


Spectra change during electron irradiation

Band no.	1	2	3	4	5	6	7	8	9	10
Product	H <sub>2</sub> O	C <sub>2</sub> H <sub>5</sub> OH	CO <sub>2</sub>	HNCO	OCN <sup>-</sup>	CO	CH <sub>2</sub> CNH	NH <sub>3</sub>	CH <sub>3</sub> CHO	HCHO

# Destruction of EtA in mixed water/EtA ice under electron irradiation

Deposition Thickness: 1.9  $\mu\text{m}$



■ **7.98 % EtA Destroyed**  
 $2.92 \times 10^{17} \rightarrow 2.687 \times 10^{17}$

■ Half-life dose:  
**5.1 eV/16u**



# Astrochemical Implications

Estimated half life of EtA ice in dense clouds and cometary ices

(only for order-of-magnitude estimates)

Location	Lifetime of ices (yr)	Depth (cm)	Dose rate (eV molecule <sup>-1</sup> yr <sup>-1</sup> )	Half-life (yr)	Half-life diluted in water (yr)
KBO	$4.6 \times 10^9$	...	$5.6 \times 10^{-3}$	$3 \times 10^2$	$9.1 \times 10^2$
(40 au beneath H <sub>2</sub> O ice)	...	$10^{-3}$	$1.6 \times 10^{-8}$	$1.06 \times 10^8$	$3.2 \times 10^8$
Cold dense cloud	$10^7$	...	$3 \times 10^{-7}$	$5.7 \times 10^6$	$1.7 \times 10^7$

- EtA ice in cold dense clouds has half life of  $5.7 \times 10^6$  years; if diluted in water, it is  $1.7 \times 10^7$  years.
- EtA ice on the Kuiper Belt Objects (KBOs) has only a half-life of 300 years. However, if it is buried below the water ice with a depth of  $10^{-3}$  cm, its half life could be  $1.06 \times 10^8$  years; if it is diluted in water, its half life could be  $3.2 \times 10^8$  years.

# SUMMARY

- The **IR spectra** of pure EtA, water-EtA mixtures under electron irradiation processes were obtained. CD changes show that it underwent an exponential decay with a half-life dose of 1.7 eV/16u, 5.1 eV/16u, respectively. The formation products include H<sub>2</sub>O, CO, CO<sub>2</sub>, OCN<sup>-</sup>, HNCO, NH<sub>3</sub>, C<sub>2</sub>H<sub>5</sub>OH, CH<sub>3</sub>CHO, etc.
- **VUV spectra** of the EtA ice under irradiation show the thin ice was destroyed, and water was formed.
- The survivability of EtA in ISM and the Solar system is analysed. The EtA ice on the Kuiper Belt Objects (KBOs) has only a half-life of 300 years, showing that it is destructed easily. However, if it is buried below the water ice with a depth of 10<sup>-3</sup> cm, its half life could be 1.06×10<sup>8</sup> years; if it is diluted in water, its half life could be 3.2×10<sup>8</sup> years. .

# ACKNOWLEDGMENTS



THE ROYAL SOCIETY



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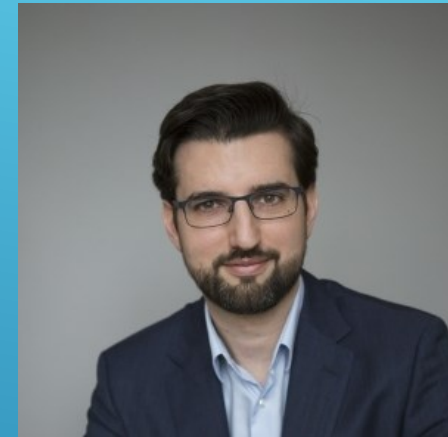
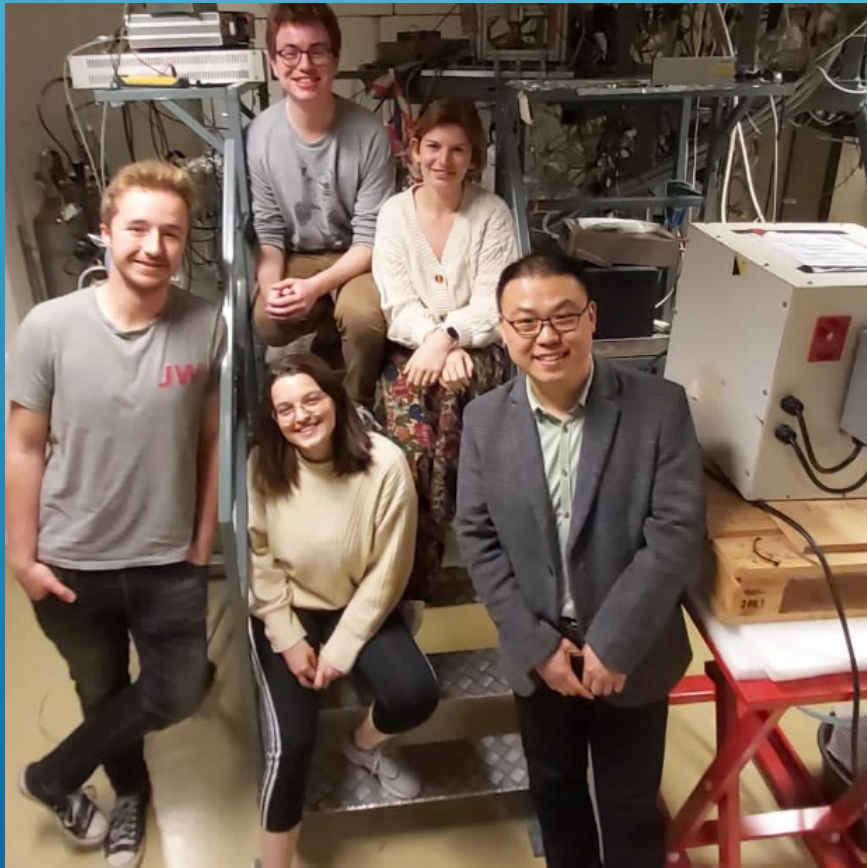
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# OUR GROUP



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**Any questions?**

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